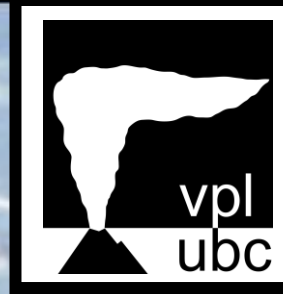




*2012 Saturday Morning Lectures Series at TRIUMF*



# **THE ASCENT of KIMBERLITE: Taxicabs for Diamonds**



**Professor Kelly Russell**

*Volcanology & Petrology Laboratory  
University of British Columbia, Vancouver*

# QUESTION:

**What Allows for Ascent of Diamondiferous Kimberlite?**

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## ORGANIZATION

**Diamond Basics**

**Mantle Basics**

**Kimberlite Basics**

**THE ISSUE**

**Answer**

**Proof of Concept**

**A NEW THEORY**

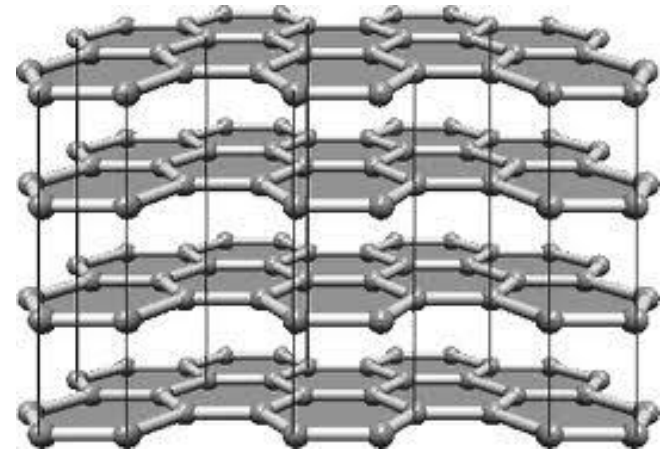


# Diamond Basics

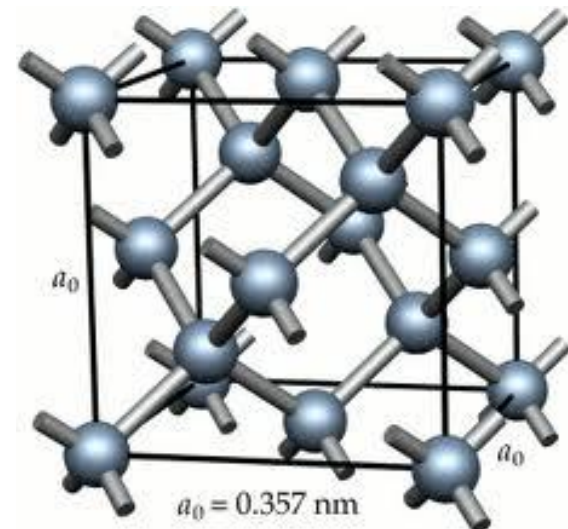
## Carbon Polymorphs (C)

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**Graphite (Hexagonal)**  
**Planar layered structure**  
**Soft Mineral 2/10**



**Diamond (Cubic)**  
**Cubic structure**  
**Hard Mineral 10/10**



# Diamond Basics

**Diamond Size (Carat = 0.2 gm : Origin Carob seed)**

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# Diamond Basics

## Diamond Size



*29 carat*



*215 carat*

# Diamond Basics

## Stability of Carbon Polymorphs (C)

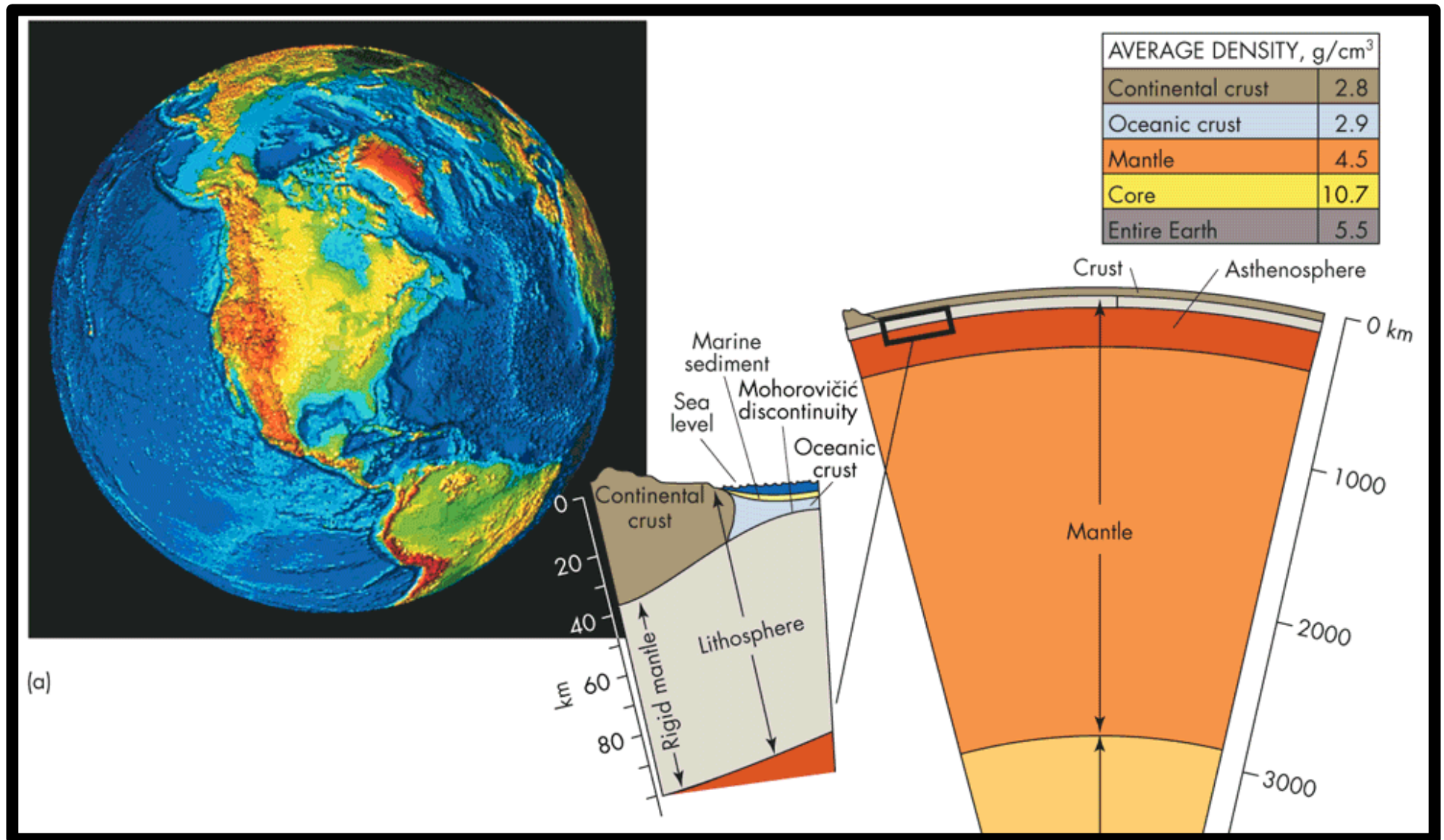
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Courtesy Maya Kopylova (2011)

# Mantle Basics

## Where are the Carbon Polymorphs (C)



# Mantle Basics

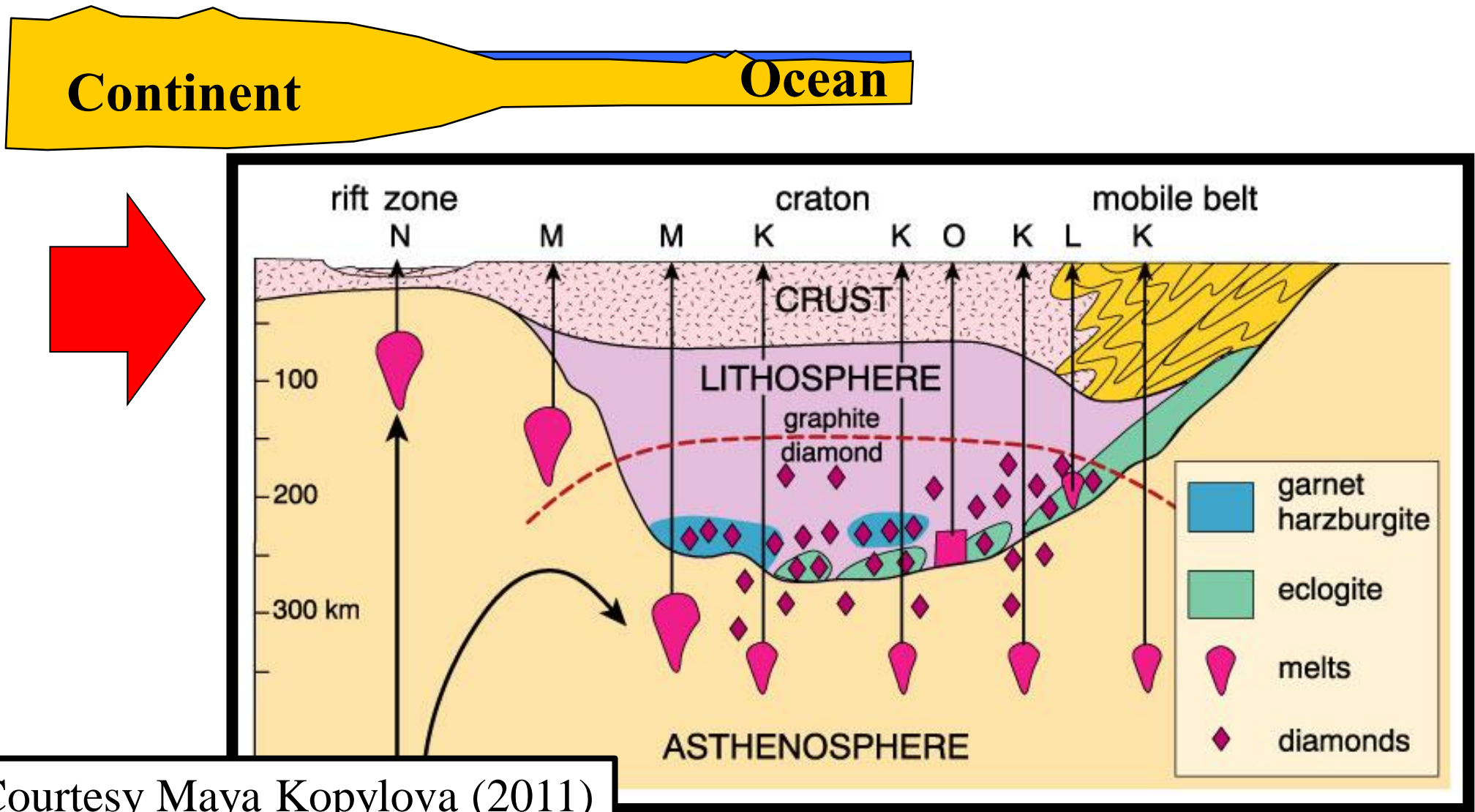
## Where are the Carbon Polymorphs (C)

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# Mantle Basics

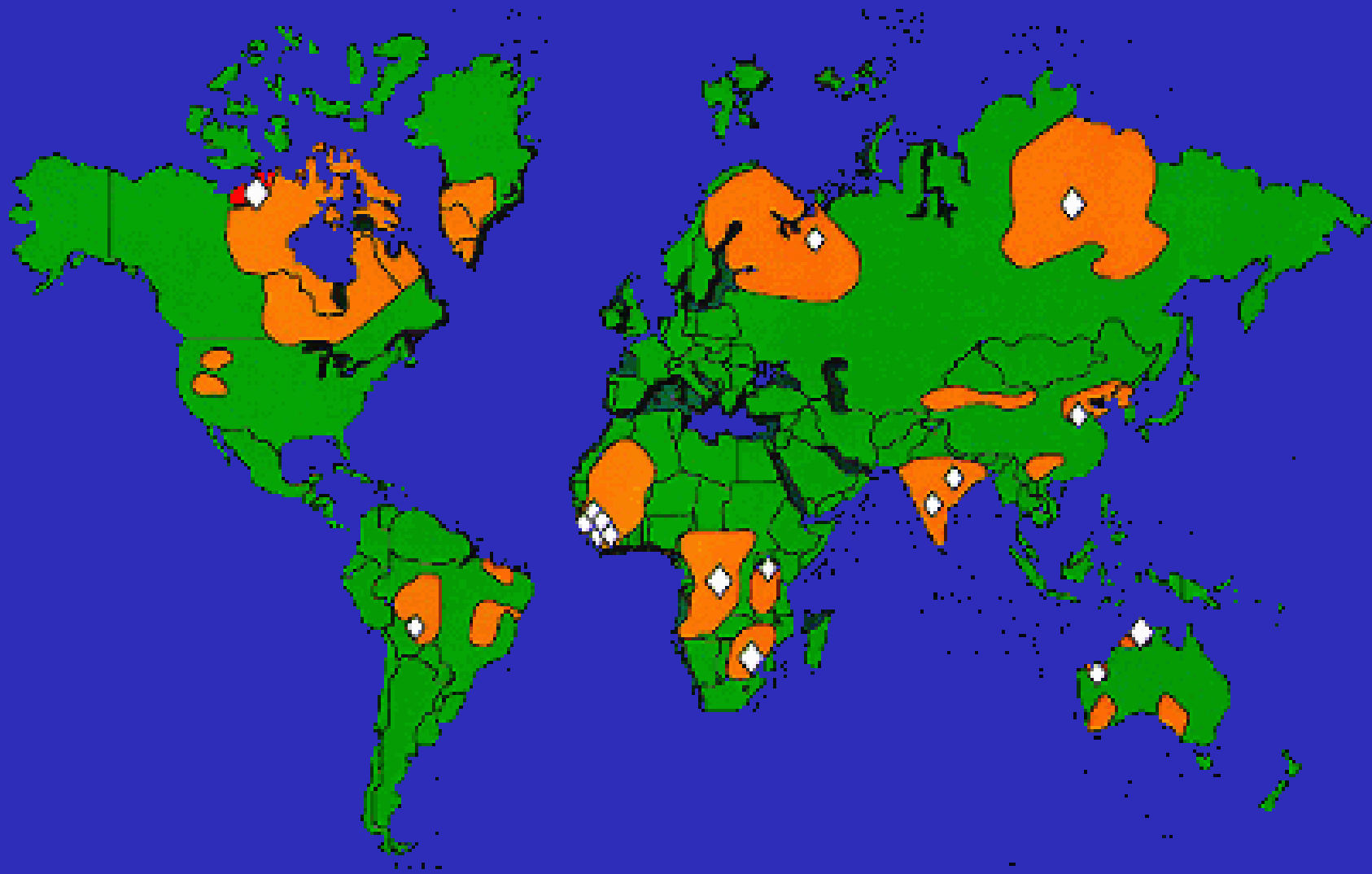
## Where are the Carbon Polymorphs (C)



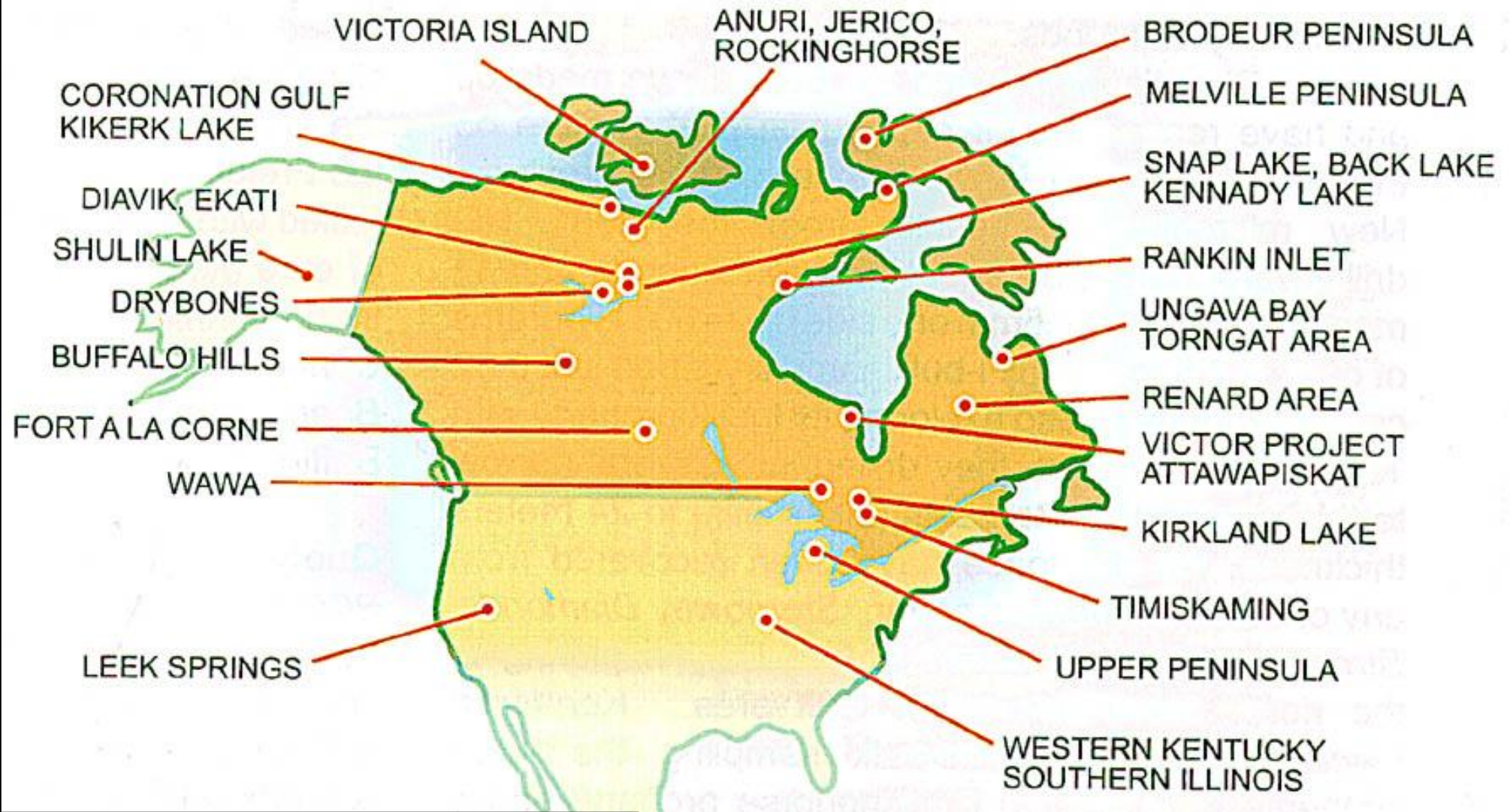
Courtesy Maya Kopylova (2011)

# Kimberlite Basics

## Worldwide Archaen Cratons and Diamond Mines

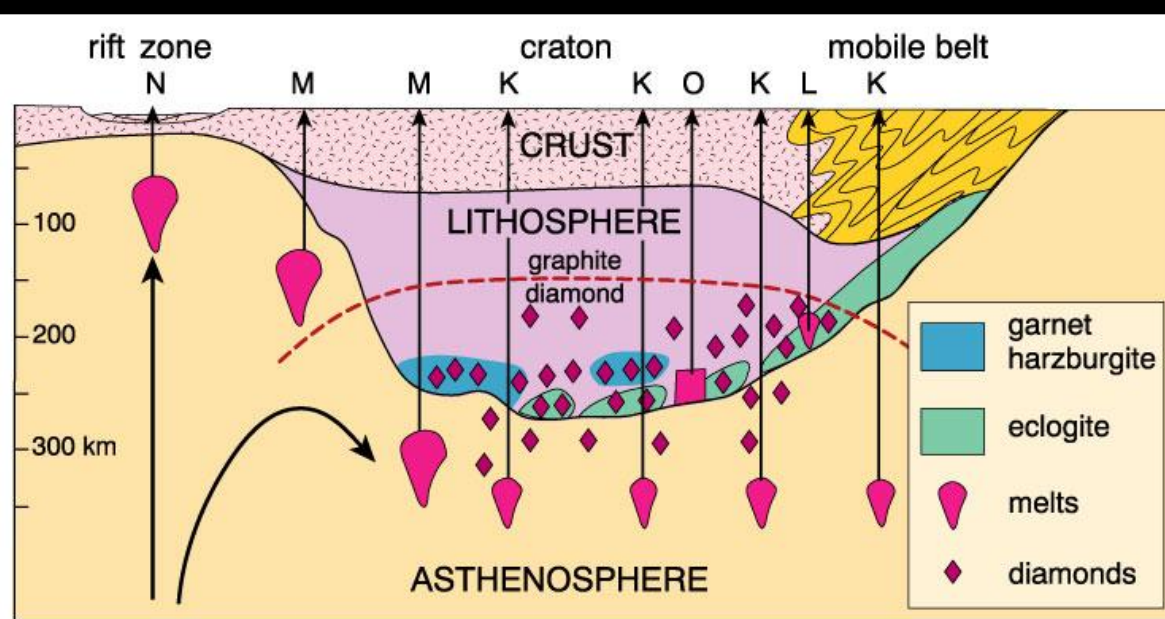
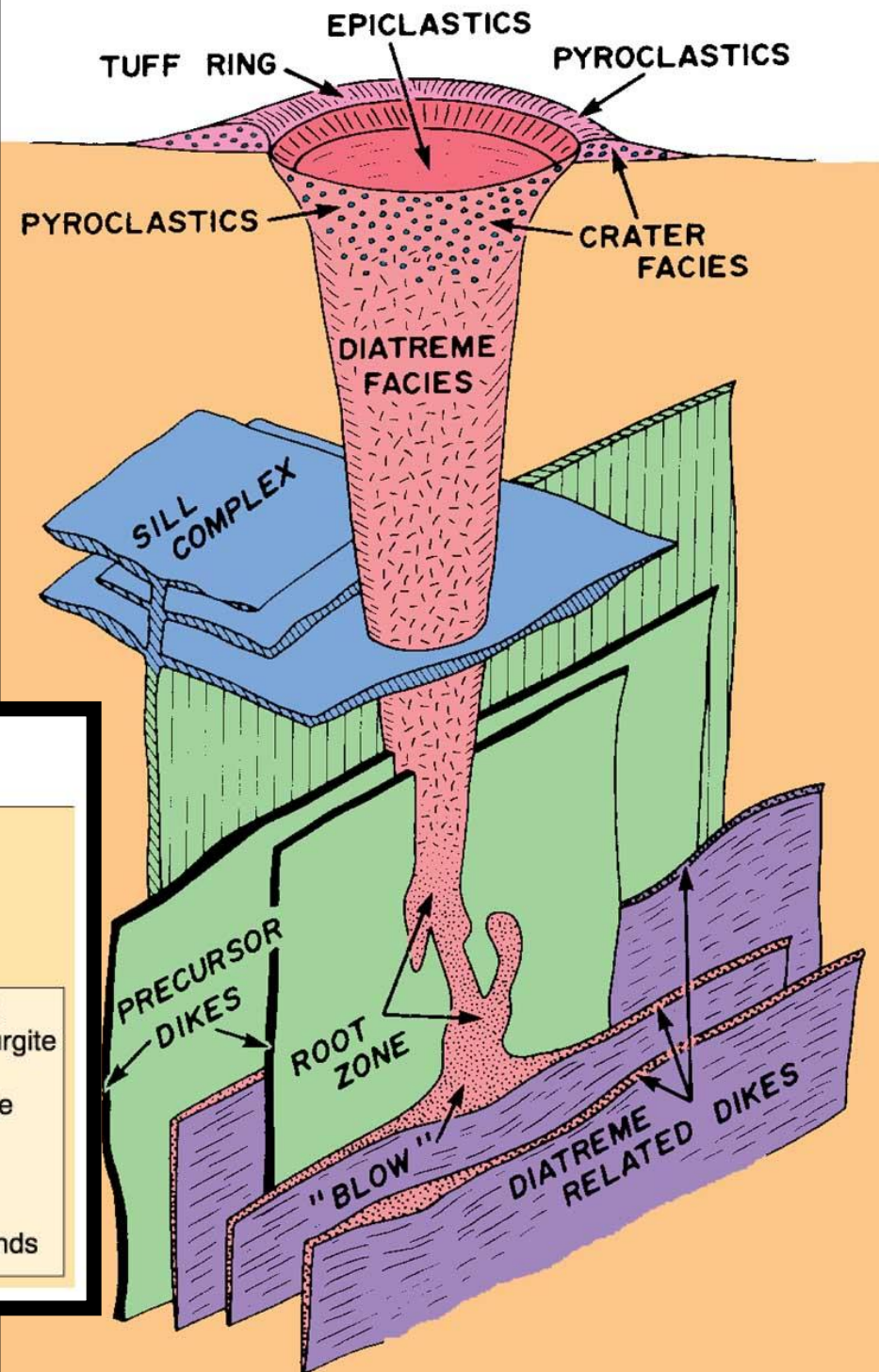


# Canadian Kimberlite Bodies & Mines



# Kimberlite Basics

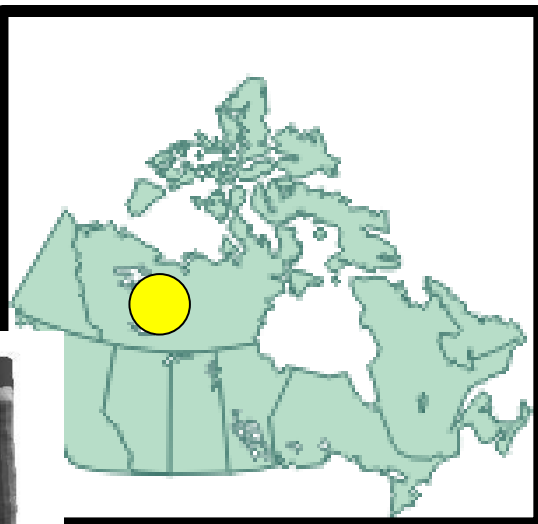
## Volcano & Pipe



# Diavik Kimberlite Mine

A418

A154N



te pipes at Diavik  
d mines, NWT

V: 6.2 MCM  
SA: 245882 m<sup>2</sup>  
D: 715 m

A154N: ~56.0 $\pm$ 0.5 Ma

A154S: ~55.5 $\pm$ 0.5 Ma

A418: ~55.2 $\pm$ 0.5 Ma

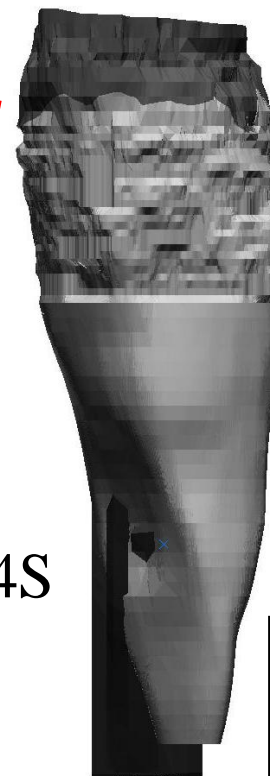
volc

A154S

5.6 MCM  
214621 m<sup>2</sup>  
500 m

1km

V: 5.9 MCM  
SA: 206327 m<sup>2</sup>  
D: 435 m





*Diavik: A154N/S*





# GOAL: A simple mechanism for kimberlite ascent

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## THE ISSUE

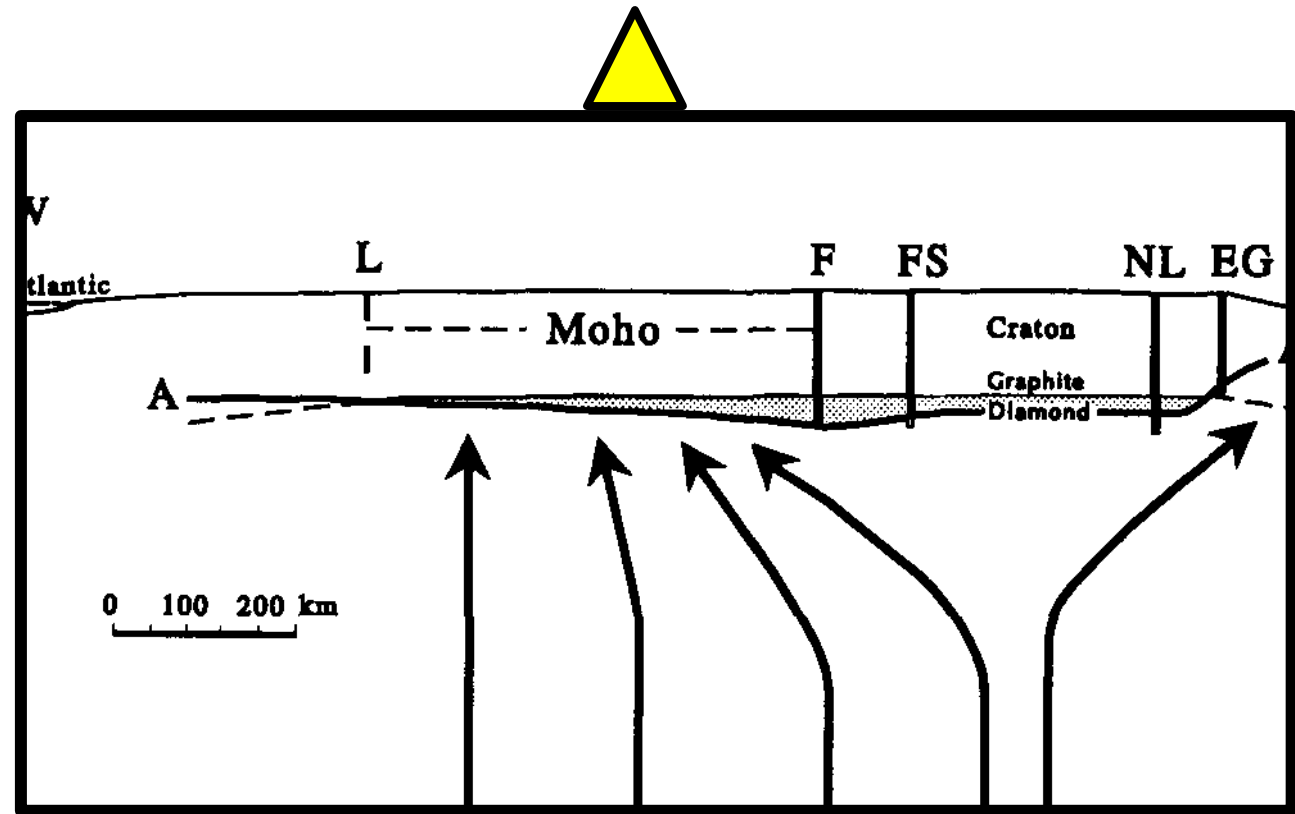
The Problem

Factoids

The Answer

Proof of Concept

A NEW THEORY



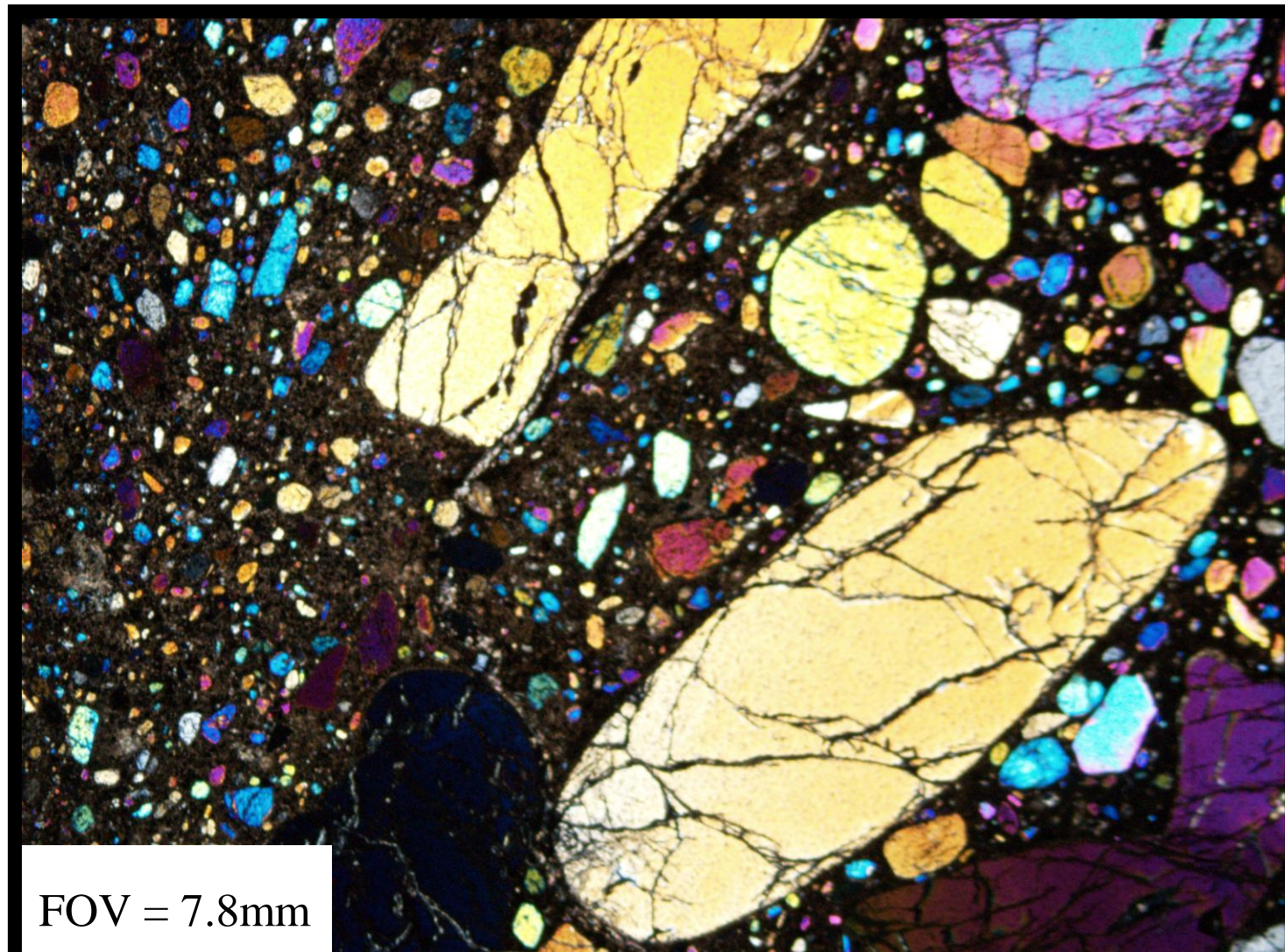
Helmstaedt & Gurney, 1995

# BASIC FACTOIDS :

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**1 ) The primary composition of kimberlite melts especially volatile contents, remains unmeasured.**

- ☐ Volatile enriched
- ☐ Low  $\eta$
- ☐ Fragile  
(poor glass formers)
- ☐ High Reactivity

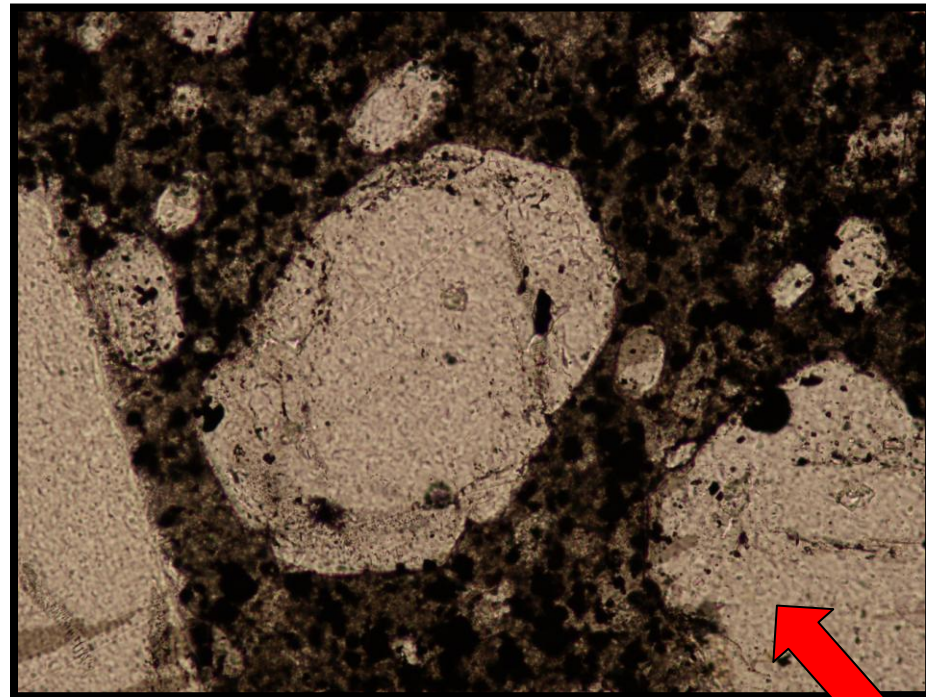


FOV = 7.8mm

# BASIC FACTOIDS:

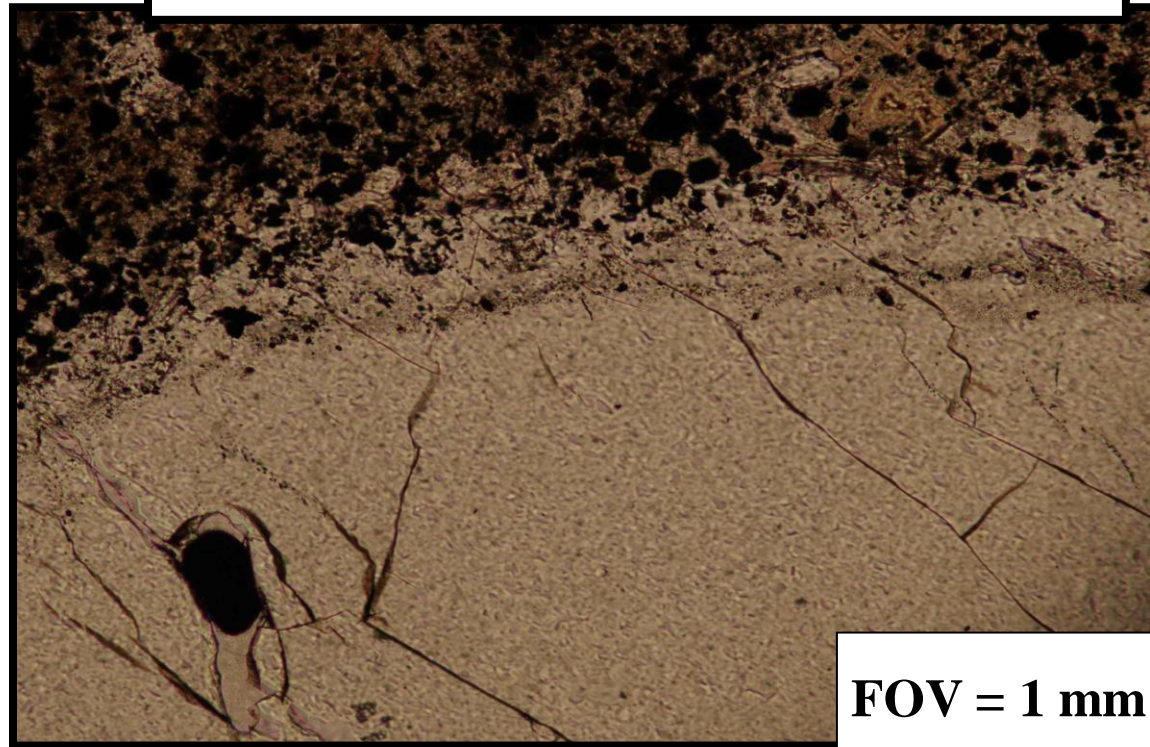
2 ) Olivine xenocrysts show early reduction in size and rounding;  
but later olivine crystallization is expressed as overgrowths.

FOV = 1 mm



*Brett & Russell (2009)*

*Macrocryst: Core and jacket*

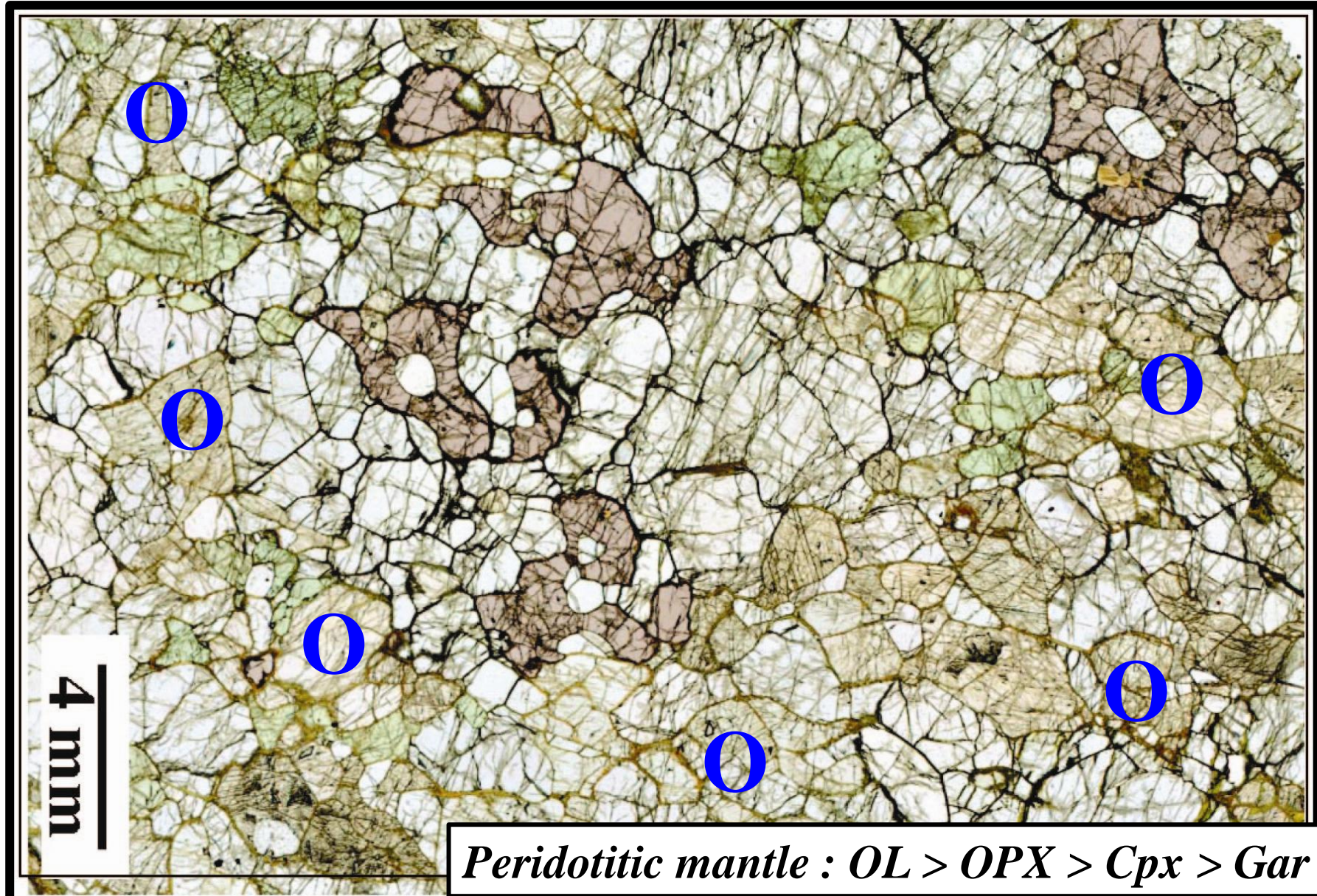


FOV = 1 mm

**ROUND Core  
EUHEDRAL Jacket**

# BASIC FACTOIDS:

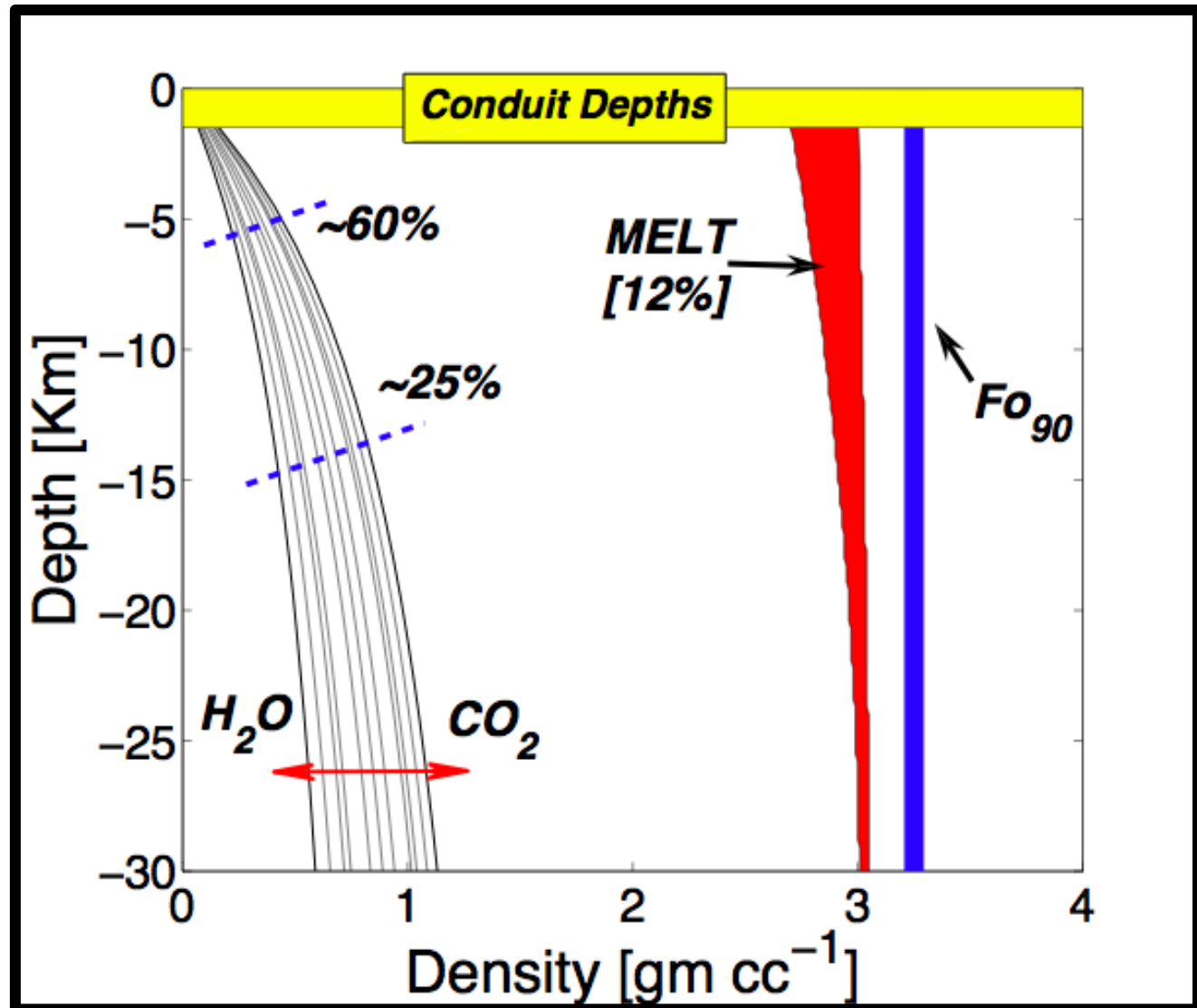
3 ) Kimberlite contains abundant xenocrysts of Ol, Cpx, Gar, ... **BUT .. ORTHOPYROXENE** is rare



# BASIC FACTOIDS :

**4 ) Kimberlite magmas have high solids content ( $> 25\%$ ) and ascend through  $> 200$  km of mantle lithosphere.**

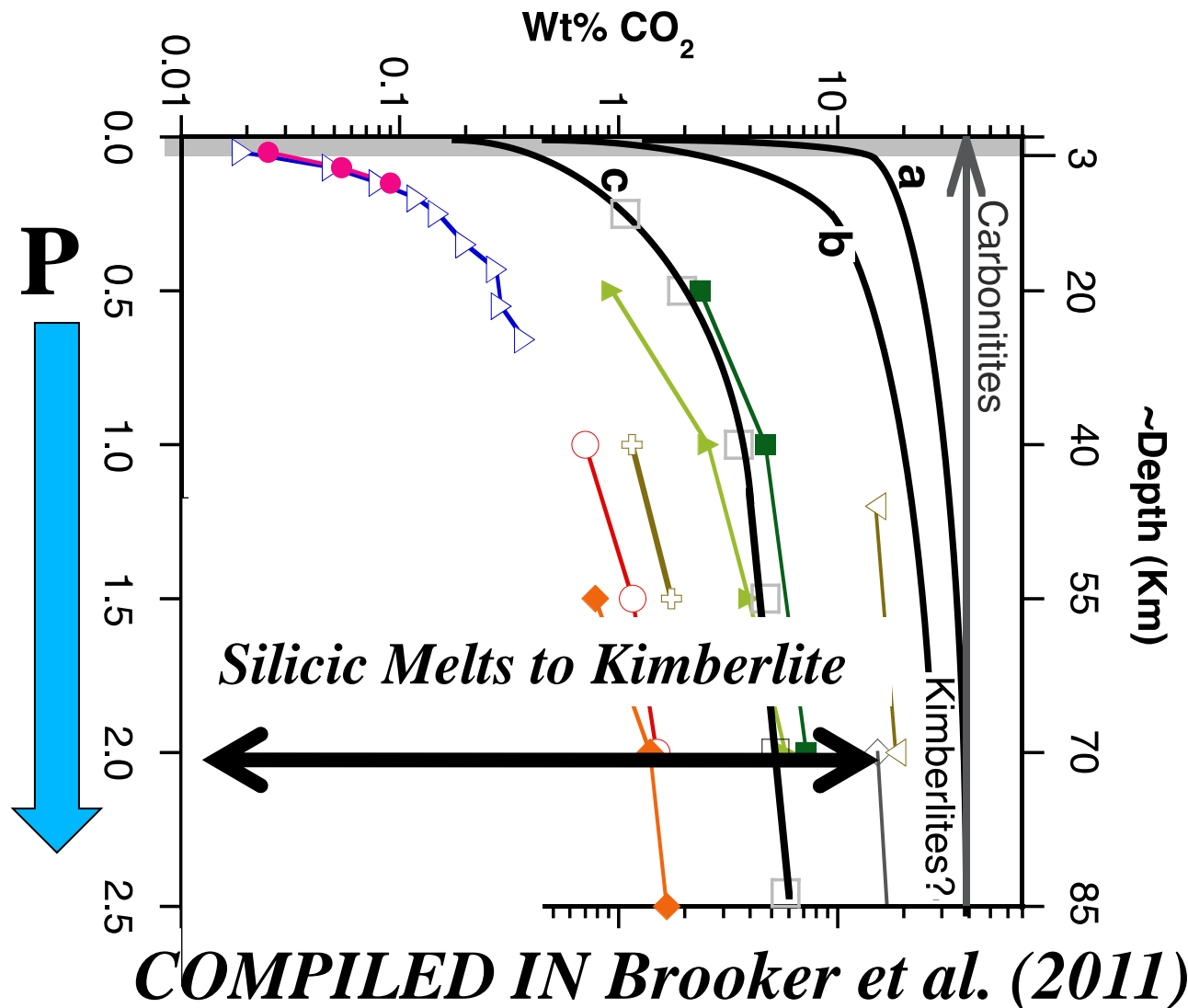
- ☐ Ascent is fast
- ☐ Ascent is continuous
- ☐ Buoyancy = Volatiles
- ☐ Deep seated Volatiles



# BASIC FACTOIDS:

**5 )  $\text{CO}_2$  (+  $\text{H}_2\text{O}$ ) solubility in silicic magmas is limited; precludes extraordinary sequestration of volatiles in melt.**

- Volatiles for Buoyancy
- W & H 2007 Model
- Exsolution > 2 GPa
- Fluid-filled crack tips
- 120 km Cracks?



# **Kimberlite - Ascent**

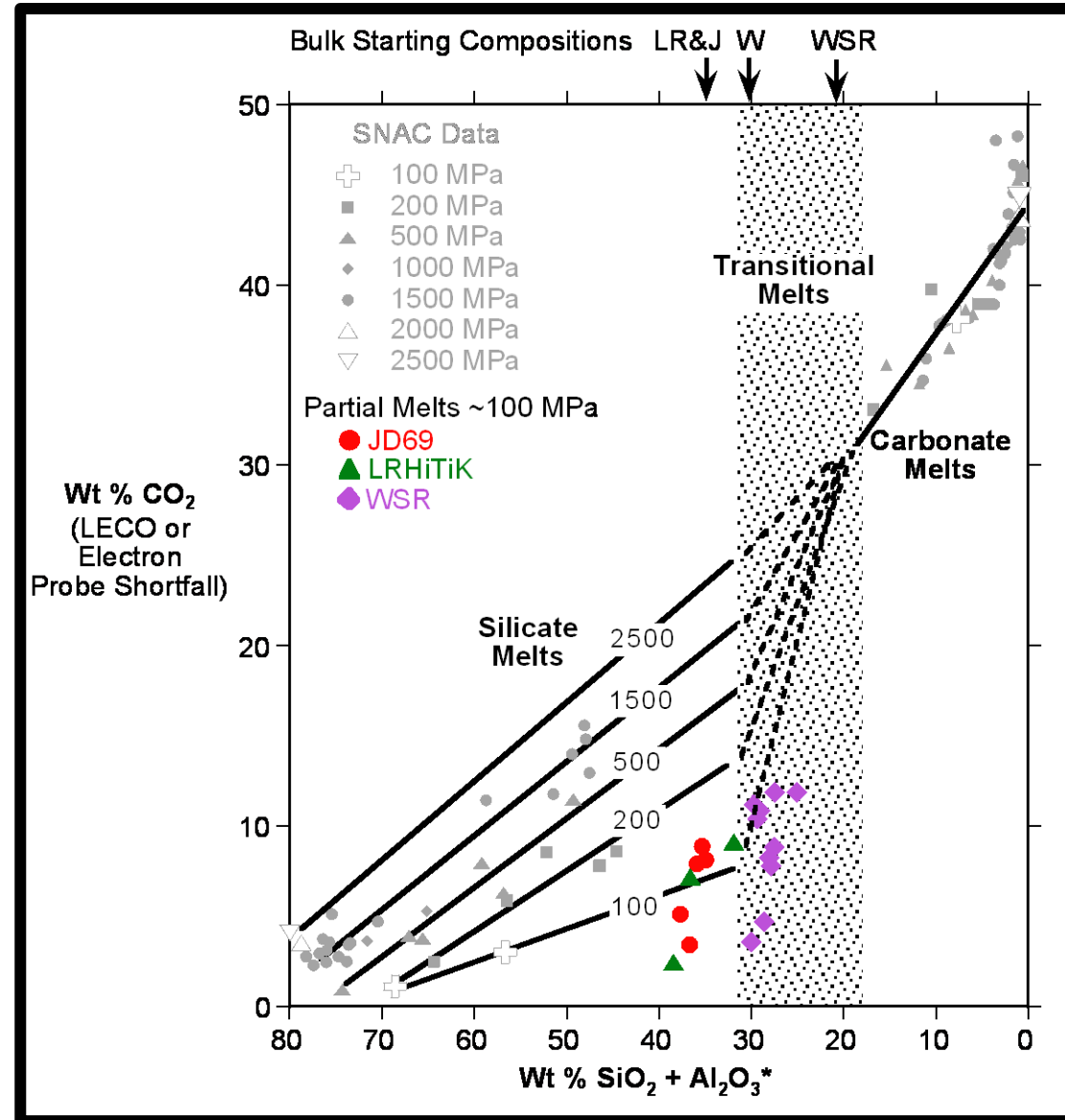
- **Volatile contents need to be HIGH (at depth)**
- **How HIGH?**
- **High enough to induce vesiculation within mantle at the greatest depths of sampling (see geotherms)**
- **High enough to support rapid ascent**
  - **preserves diamond**
  - **carries many & large xenoliths (ol +opx+cpx+gar)**
  - **mechanically mills xenoliths & megacrysts**
  - **no early crystallization (overgrowths & phenos?)**



*HOW?*

# ANOTHER FACTOID:

6 ) CO<sub>2</sub> solubility in carbonate melts is limited only by melt stoichiometry; Na-carbonate melt > 40% dissolved CO<sub>2</sub>

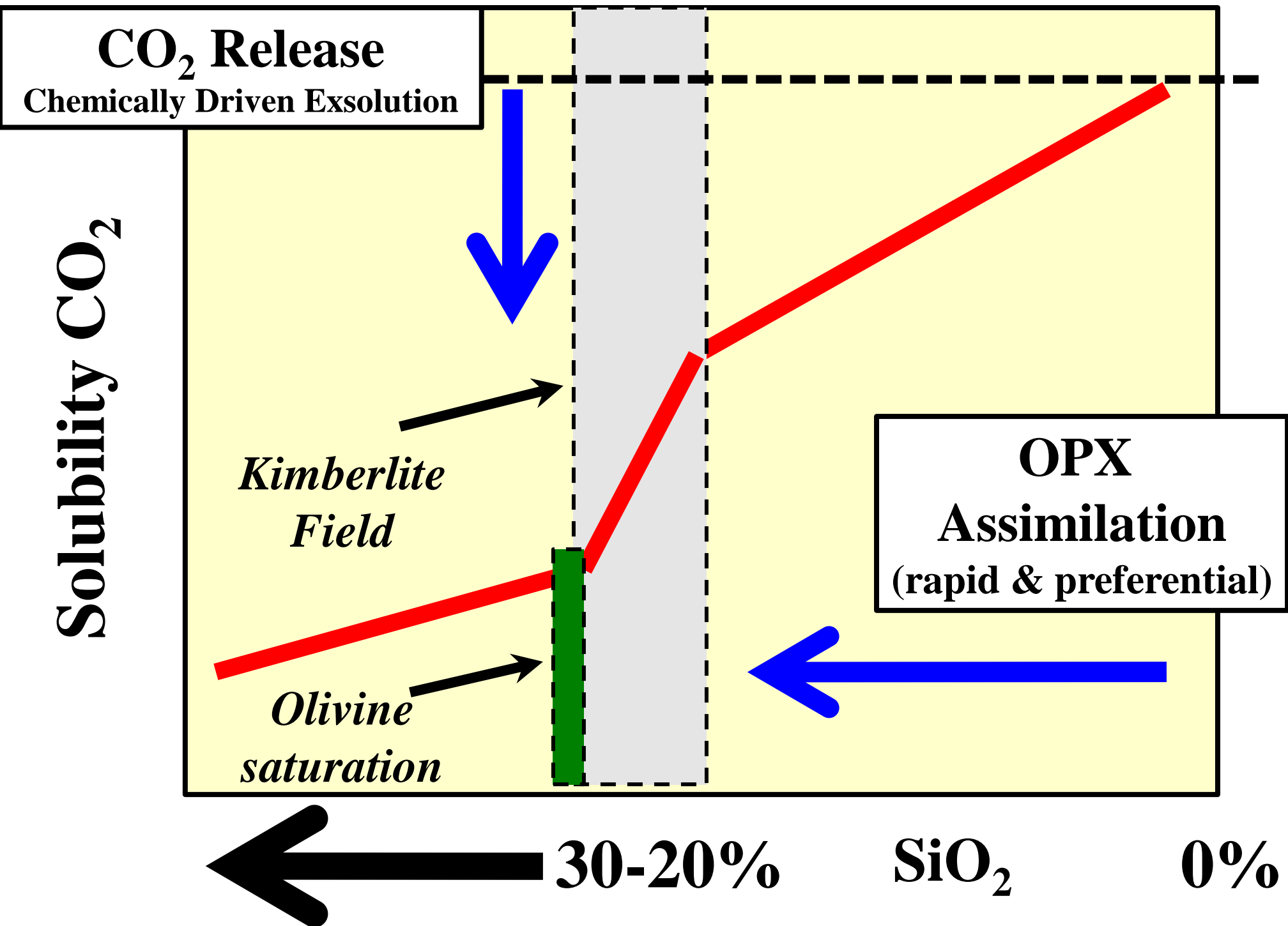


*Brooker et al. (2011)*

# THE INDUCTIVE IDEA

## Assimilation-Induced Foaming The Mechanism for Kimberlite Ascent

- **ALL kimberlites start as carbonatic melts**
- **Carbonatic melts have stoichiometric CO<sub>2</sub> contents**
- **Melts enter and sample cratonic mantle lithosphere**
- **Peridotite disaggregates - OPX dissolves preferentially**
- **Carbonate melt + MgSiO<sub>3</sub> = EFFERVESCENCE**

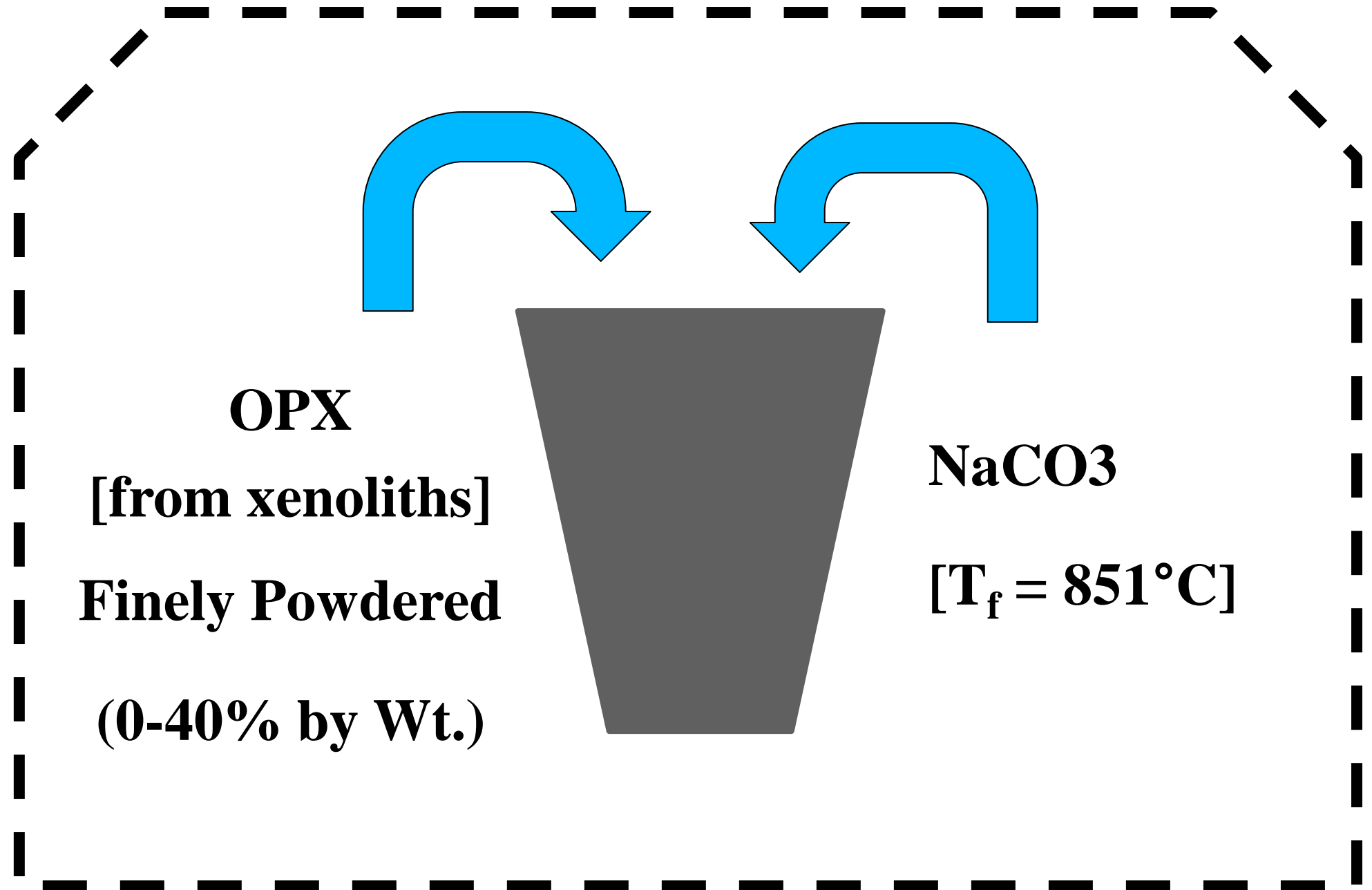


# DEDUCTIVE TESTS OF IDEA



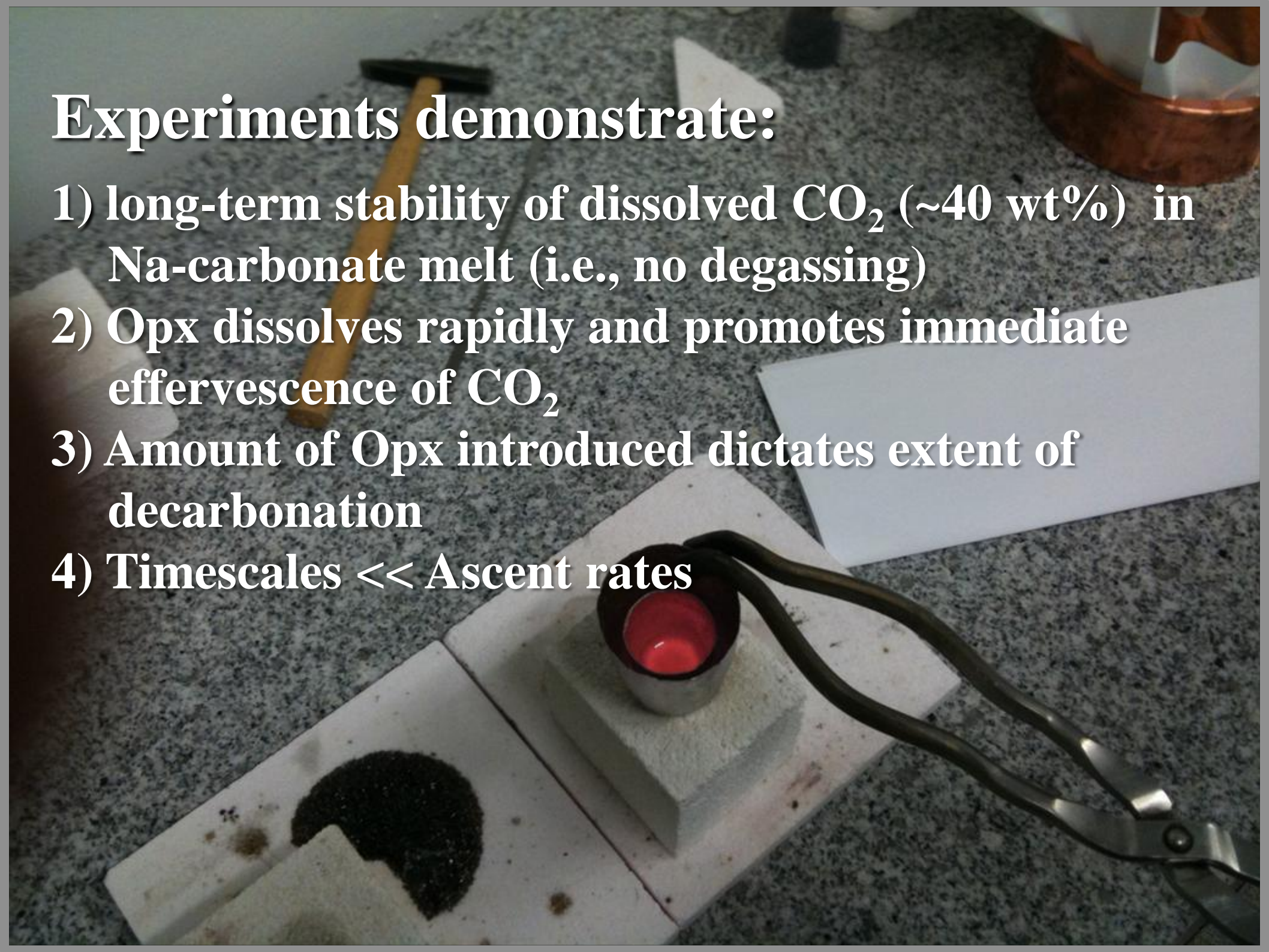
# NEW HIGH-T WEIGHT LOSS EXPERIMENTS

Super-liquidus Conditions at  $T > 1050^{\circ}\text{C}$  for  $> 1\text{hour}$

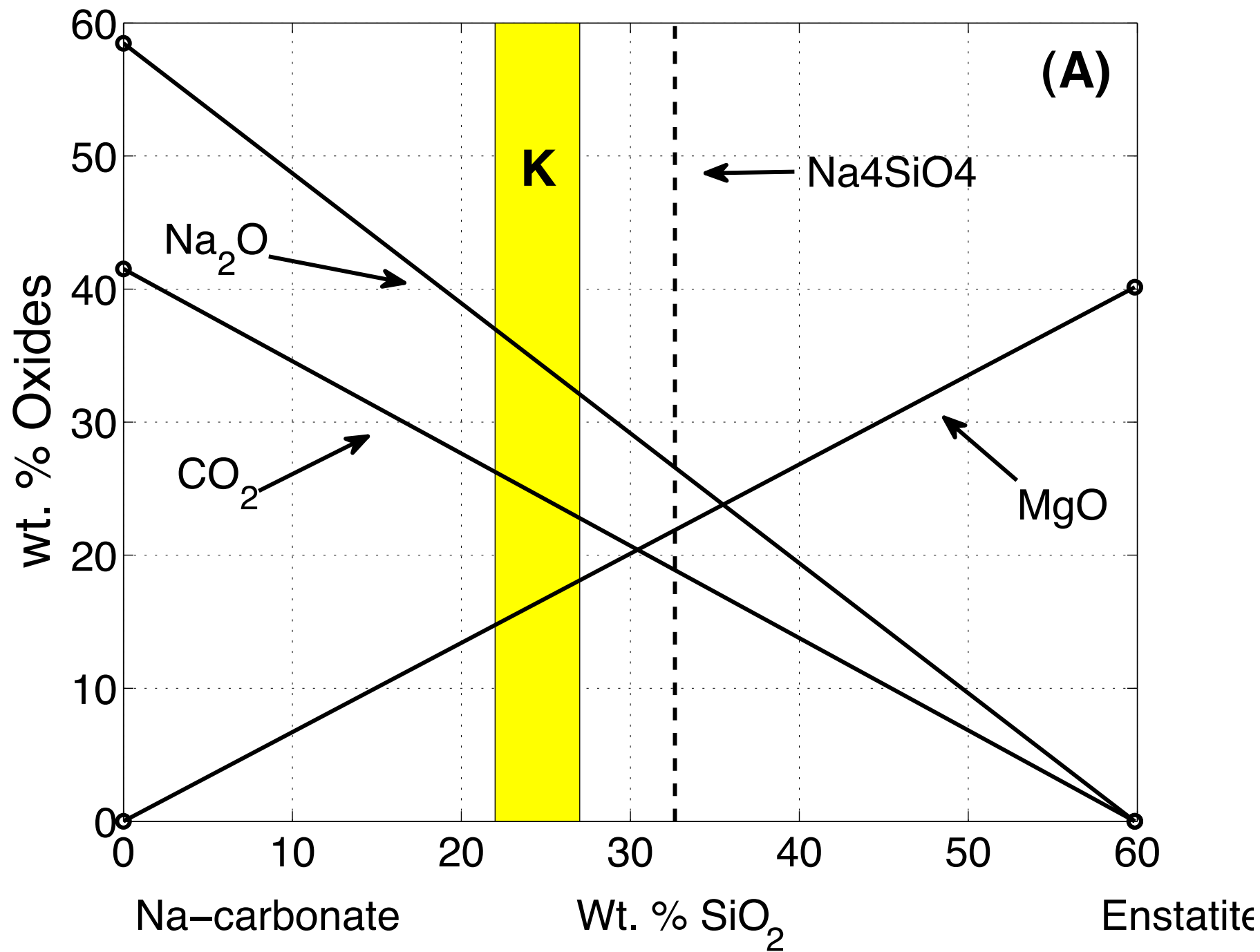


# Experiments demonstrate:

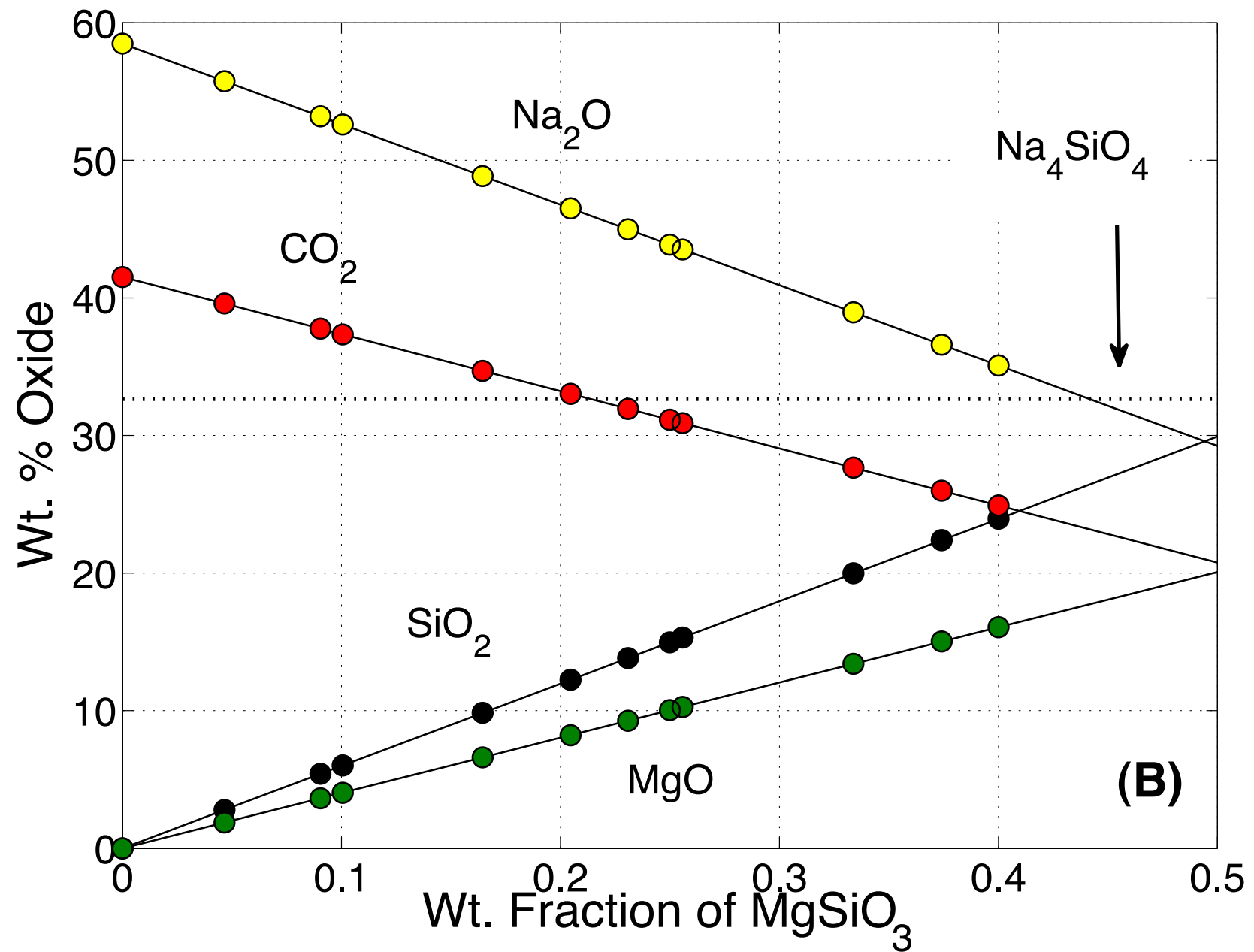
- 1) long-term stability of dissolved  $\text{CO}_2$  (~40 wt%) in Na-carbonate melt (i.e., no degassing)
- 2) Opx dissolves rapidly and promotes immediate effervescence of  $\text{CO}_2$
- 3) Amount of Opx introduced dictates extent of decarbonation
- 4) Timescales  $\ll$  Ascent rates



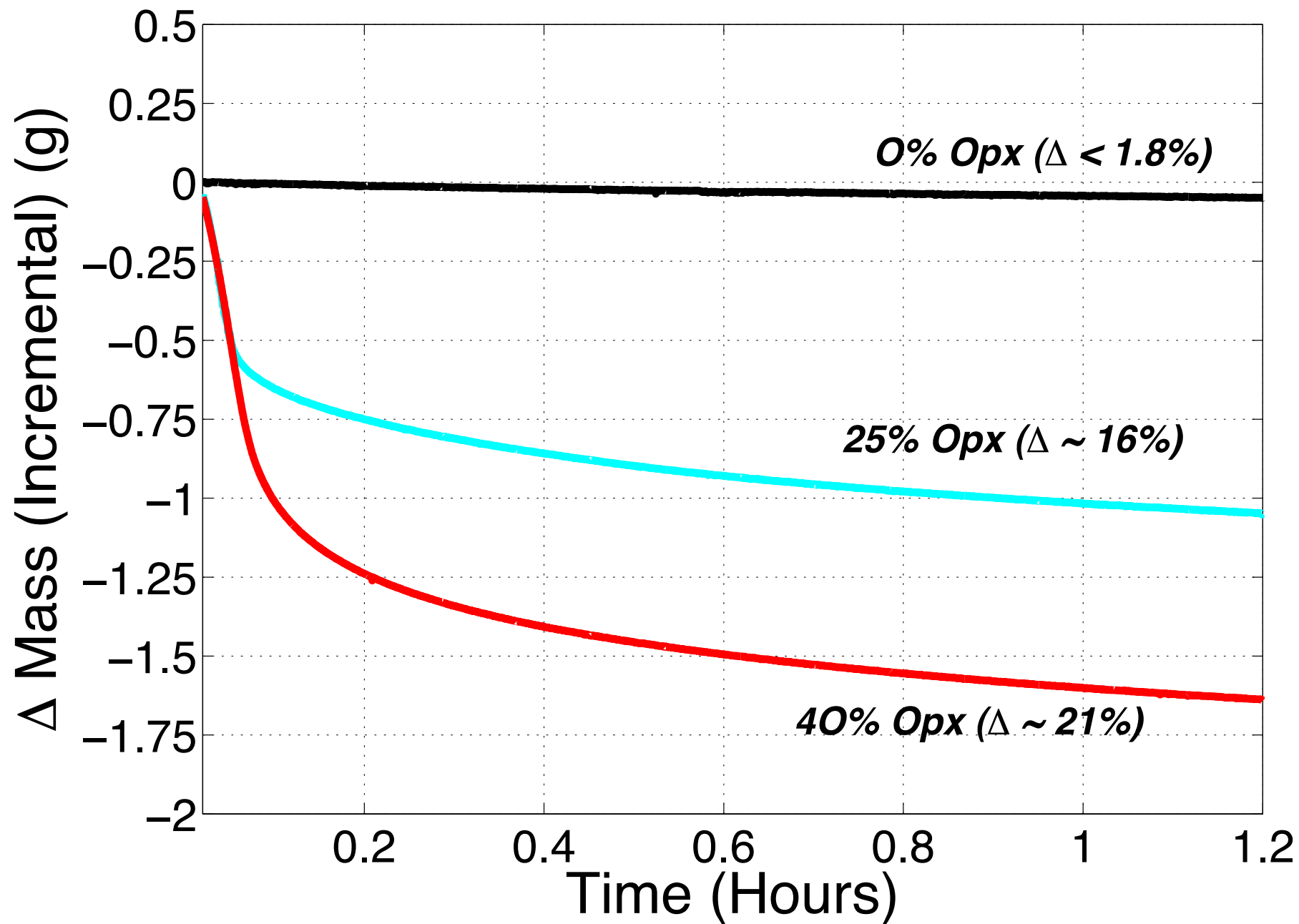
# MECHANICAL MIXTURES



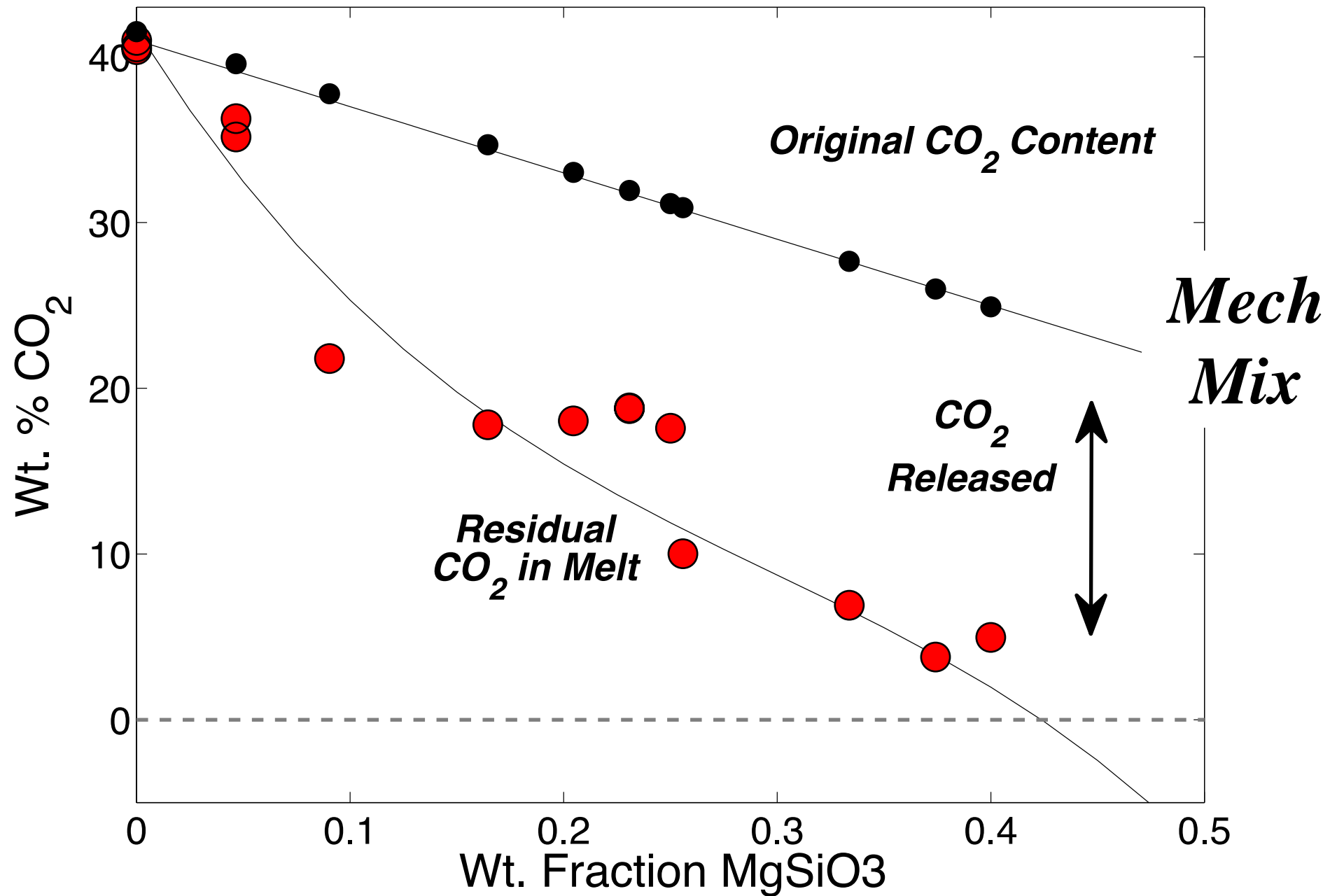
# MECHANICAL MIXTURES



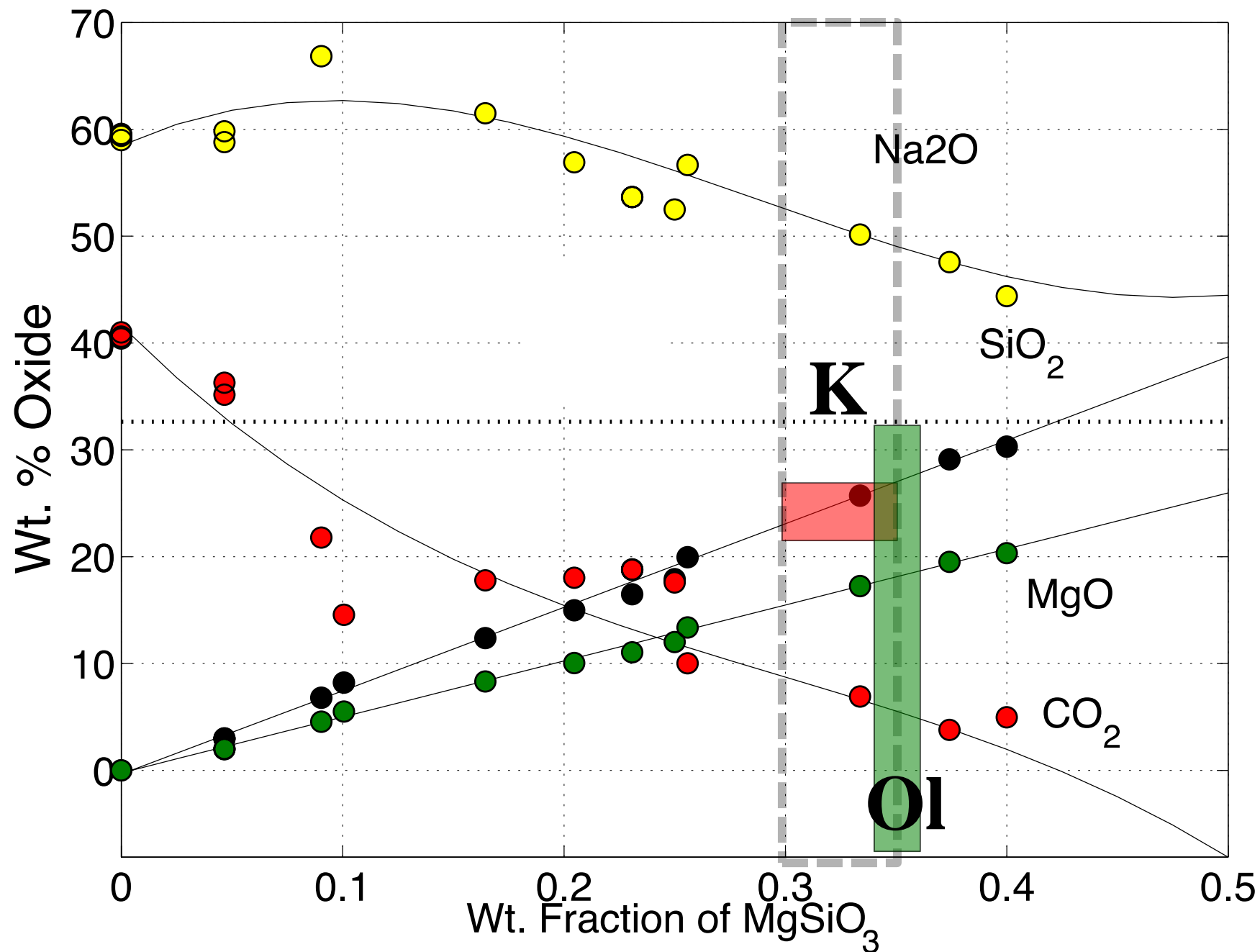
# RESULTS: Transient Experiments



# RESULTS: Decarbonation Experiments

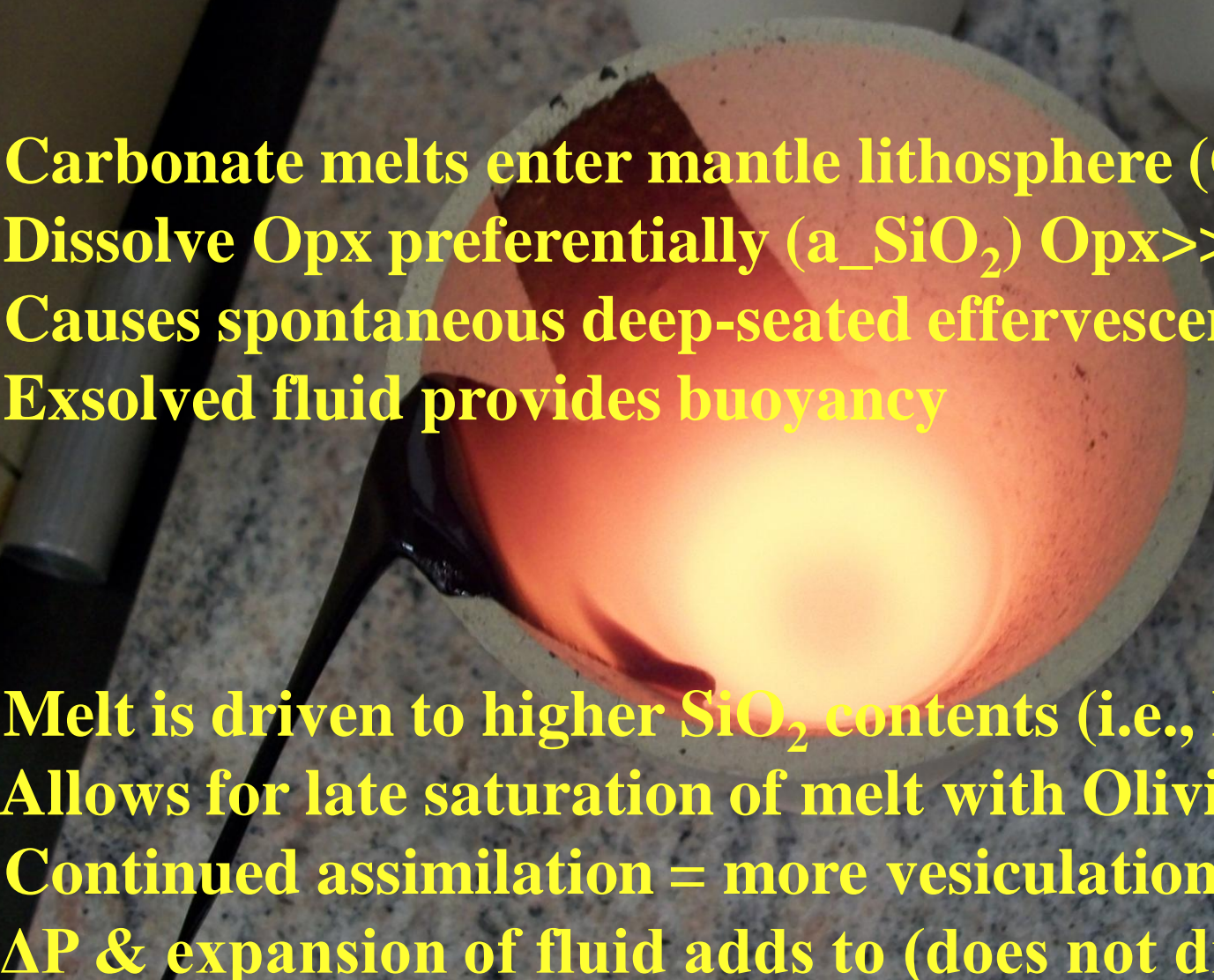


# RESULTS: Melts from Chemical Mixtures



# **Kimberlite Ascent (THE ANSWER – FINALLY)**

## **Kimberlite: Assimilation-Fueled Ascent of Carbonatite**

- 
- 1) Carbonate melts enter mantle lithosphere (OPX-rich)
  - 2) Dissolve Opx preferentially ( $a_{\text{SiO}_2}$ )  $\text{Opx} \gg \text{Cpx}$ , Gar, Ol
  - 3) Causes spontaneous deep-seated effervescence
  - 4) Exsolved fluid provides buoyancy
  - 5) Melt is driven to higher  $\text{SiO}_2$  contents (i.e., Kimberlite)
  - 6) Allows for late saturation of melt with Olivine
  - 7) Continued assimilation = more vesiculation = more cargo
  - 8)  $\Delta P$  & expansion of fluid adds to (does not drive) buoyancy